Nickel-catalyzed [3 + 2] Cycloaddition of α , β -Unsaturated Ketones with Vinyl Oxiranes

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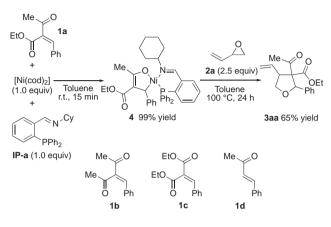
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A nickel-catalyzed cycloaddition of α,β -unsaturated ketones with vinyl oxiranes to afford polysubstituted tetrahydrofurans has been developed. The key intermediate is a five-membered nickelacycle, which is formed by oxidative cyclization of nickel(0) to α,β -unsaturated ketones.

Transition-metal-catalyzed cycloadditions represent a powerful strategy of assembling complex heterocyclic compounds from readily available starting materials. Thus, the development of new formal reaction patterns still constitutes an important ongoing challenge.¹ Among these, a formal [3 + 2] cycloaddition of oxiranes with olefins seems to be an attractive strategy to construct tetrahydrofuran frameworks in a single operation. However, these types of cycloaddition are rarely reported.² We recently developed a series of nickel-catalyzed cycloaddition using α -ester-substituted α,β -unsaturated ketones as a key component.³ During the course of our study, we found that α -ester-substituted α,β -unsaturated ketones 1 reacted with vinyl oxiranes 2 to furnish polysubstituted tetrahydrofuran 3 via a formal [3 + 2] cycloaddition; the reaction process included oxidative cyclization of α -ester-substituted α,β -unsaturated ketones to nickel(0).

We first examined the stoichiometric reaction. The reaction of α -ester-substituted α,β -unsaturated ketone 1a was treated with $[Ni(cod)_2]$ and iminophosphine ligand⁴ **IP-a** in toluene at ambient temperature for 15 min (Scheme 1). Oxa-nickelacycle 4 was obtained in quantitative yield.^{3,5} Neither benzylidenesubstituted pentane-2,4-dione 1b, diethyl malonate 1c, nor enone 1d participated in oxidative cyclization with [Ni(cod)₂]/IP-a, and each was recovered unchanged quantitatively. Treatment of 4 with vinyl oxirane (2a) in toluene at 100 °C for 24 h afforded tetrahydrofuran 3aa in 65% yield. The results prompted us to investigate the possibilities of the catalytic reactions of 1a with 2a. Indeed, the reaction of 1a with 2a in the presence of 10 mol % of [Ni(cod)₂] and 20 mol % of **IP-a** in toluene at 100 °C for 24 h furnished the tetrahydrofuran 3aa in 56% yield (Table 1, Entry 1). Not even trace amounts of cycloadduct 3aa were obtained in the absence of nickel. Further examination of reaction conditions revealed that the cycloaddition proceeds at lower reaction temperature (60 °C), providing a higher yield of desired cycloadduct 3aa (75% yield, Entry 2). It was also found that the reaction furnished 3aa in 98% yield at 30 °C (Entry 3). In other reaction media, the yields of 3aa were decreased (Entries 4-6). Among the iminophosphines examined, cyclohexylamine-derived ligand IP-a gave the best result (Entries 3 and 7-9). With other phosphine ligands, such as PMe₃, PMe₂Ph, and PPh₃, the reaction afforded even lower amounts of desired cycloadduct 3aa (Entries 10-12).

The scope of α , β -unsaturated ketones **1** and vinyl oxiranes **2** in the cycloaddition is summarized in Table 2. It was found



Scheme 1. Stoichiometric reaction.

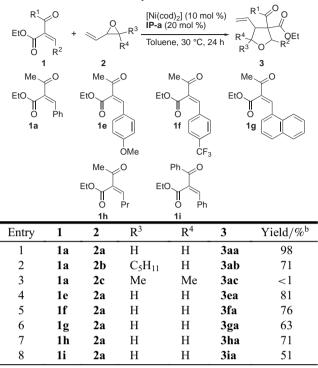
Table 1. [3 + 2] Cycloaddition of α, β -unsaturated ketone 1a with vinyl oxirane 2a with nickel catalyst^a

$He \xrightarrow{O} Ph + \underbrace{Ph}_{1a} Ph + \underbrace{Ph}_{2a} \xrightarrow{[Ni(cod)_2] (10 \text{ mol }\%)}_{\text{Toluene, 24 h}} \xrightarrow{Me}_{O} \xrightarrow{O} Ph + \underbrace{Ph}_{O} \xrightarrow{O} Ph$				
PPh ₂ Ph PPh ₂ PPh ₂ PPh ₂				N Ph
Entres	IP-b	IP-c	IP-d	V: 14 /0/ b
Entry	Ligand	Solvent	Temperature/°C	Yield/% ^b
1	IP-a	Toluene	100	56
2	IP-a	Toluene	60	75
3	IP-a	Toluene	30	98
4	IP-a	THF	30	77
5	IP-a	MeCN	30	37
6	IP-a	Pyridine	30	18
7	IP-b	Toluene	30	34
8	IP-c	Toluene	30	11
9	IP-d	Toluene	30	59
10	PMe ₃	Toluene	30	<1
11	PMe ₂ Ph	Toluene	30	<1
12	PPh ₃	Toluene	30	9

^aReactions were carried out using $[Ni(cod)_2]$ (10 mol%), ligand (20 mol%), **1a** (0.5 mmol), and **2** (1.25 mmol) for 24 h. ^bReactions afford tetrahydrofurans **3** as a mixture of inseparable diastereoisomers.

that vinyl oxirane **2b** possessing pentyl substituent on the oxirane ring also reacted with α,β -unsaturated ketone **1a** to furnish the corresponding tetrahydrofuran derivative **3ab** in 71% yield (Entry 2). However, the reaction of **1a** with vinyl oxirane **2c** possessing *gem*-dimethyl substituents on the oxirane ring did

Table 2. Nickel-catalyzed [3 + 2] cycloaddition of α,β -un-saturated ketones 1 with vinyl oxiranes 2^{a}

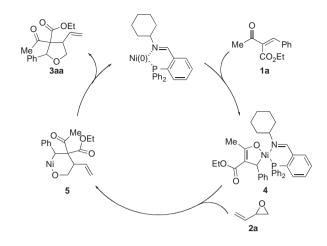


^aReactions were carried out using $[Ni(cod)_2]$ (10 mol %), **IP-a** (20 mol %), **1** (0.5 mmol), and **2** (1.25 mmol) in toluene at 30 °C for 24 h. ^bReactions afford tetrahydrofurans **3** as a mixture of inseparable diastereoisomers.

not afford the desired cycloadduct (Entry 3). A range of α,β -unsaturated ketones 1 was also tested in the reaction with 2a. β -Aryl-substituted enones possessing an electron-donating or -withdrawing group on the phenyl ring also provided corresponding cycloadducts 3ea and 3fa in 81% and 76% yields respectively (Entries 4 and 5). β -Naphthyl-substituted enone 1g also participated in the reaction with 2a to give tetrahydrofuran 3ga in 63% yield (Entry 6). The reaction of β -alkyl-substituted enone 1h provided the cycloadduct 3ha in 71% yield (Entry 7). Enone 1i reacted with 2a to furnish tetrahydrofuran 3ia in 51% yield (Entry 8).⁶

The formation of polysubstituted tetrahydrofuran **3aa** can be rationalized as arising from oxidative cyclization of nickel(0) to an enone **1a** to form *oxa*-nickelacycle **4** (Scheme 2). Subsequent coordination of vinyl oxirane **2a** and nucleophilic addition of nickel enolate to vinyl oxirane **2a** take place to afford sixmembered *oxa*-nickelacycle **5**,^{5d,7} which undergoes reductive elimination to give **3aa**; the starting nickel(0) complex is regenerated.

In conclusion, we have developed a new nickel-catalyzed [3 + 2] cycloaddition of α,β -unsaturated ketones with vinyl oxiranes to provide polysubstituted tetrahydrofurans. We manifested that the reaction is initiated by oxidative cyclization of α,β -unsaturated ketones with nickel(0). Further efforts to control the diastereoselectivity of the reaction are now in progress.⁸



Scheme 2. Plausible reaction mechanism.

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